

Total synthesis of natural (+)-membrenone C and its 7-epimer[☆]

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Dedicated to Professor S. Chandrasekaran, IISc, Bangalore on his 60th birthday

Abstract—Highly stereoselective asymmetric total syntheses of the polypropionate marine defense substance (+)-membrenone C and its 7-epimer have been achieved. Highlights of the strategy include the utilization of a desymmetrization technique to create five contiguous chiral centres from a single bicyclic precursor.

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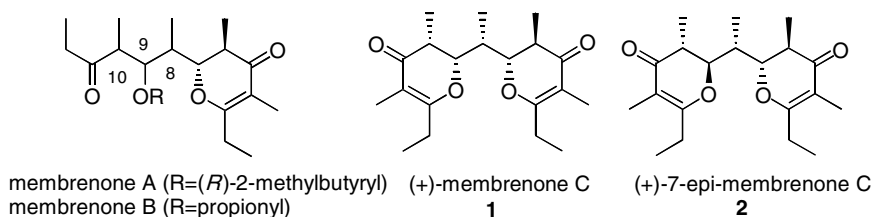
During the course of our ongoing programme of synthesis of biologically potent natural products we selected (+)-membrenone C, the polypropionate marine defense natural product as a synthetic target. Membrenone C was first isolated along with membrenone A and membrenone B from a Mediterranean mollusc *Pleurobranchus membranaceus* by Ciavatta et al.¹ Membrenones A–C were found to afford protection for the mollusc from potential predators in the hostile marine environment. Ciavatta and co-workers assigned structures to the membrenones by extensive NMR analysis, the relative and absolute stereochemistry of the natural products was confirmed by synthesis.^{2,3}

The exceptional bioactivity and extreme scarcity of the natural material together with the novel structure prompted us to attempt the total synthesis of (+)-membrenone C **1**. It was envisioned that (+)-membrenone C could be obtained from tetraketone **3** as shown in Scheme 1. The construction of **3** was to be based on a

disconnection between the C4–C5 and C11–C12 bonds via double aldol reactions as reported by Perkins and co-workers.³ The dialdehyde could be derived from a single bicyclic precursor **6** which in turn is easily synthesized, and was earlier used by our group towards the synthesis of rifamycin S,⁴ discodermolide,⁵ scytophycin C⁶ and prelactone B.⁷

The required lactone^{5,6} **7** was prepared from **6** by a protocol we had developed and utilized for a number of syntheses wherein we had exploited a desymmetrization technique to create six stereogenic centres at once.

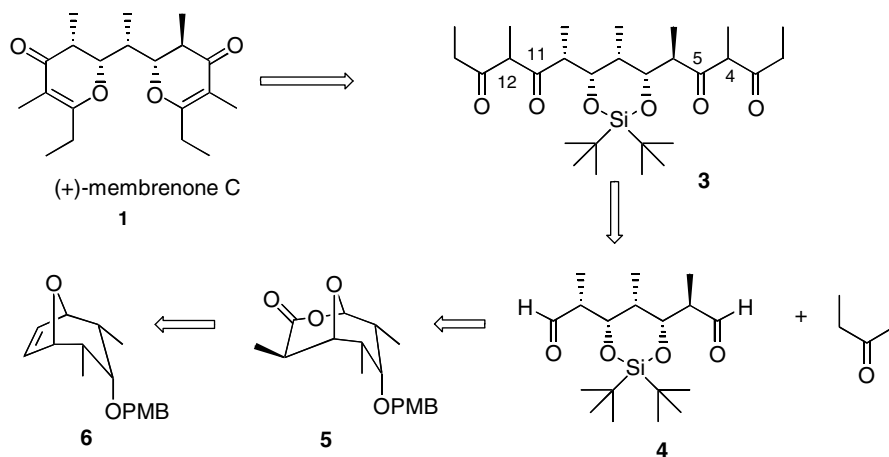
Alkylation of the lactone **7** with LHMDs/MeI afforded compound **5**⁴ as the only product which on reductive opening with LAH furnished the triol **8** (Scheme 2). The triol **8** on dibenzyl ether protection followed by PMB⁸ ether deprotection afforded diol **9**. Protection of the diol as its di-*tert*-butyl silyl ether **10** was achieved by using di-*tert*-butyl silyl bis[trifluoromethanesulfonate]



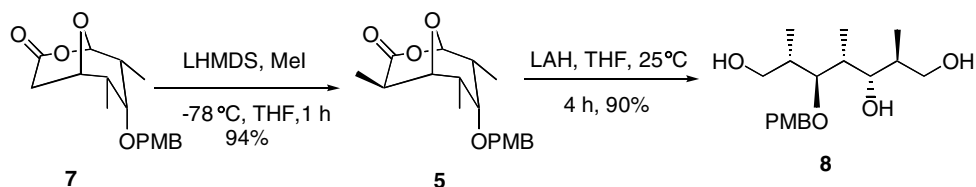
Keywords: (+)-Membrenone C and its 7-epimer; Desymmetrization technique.

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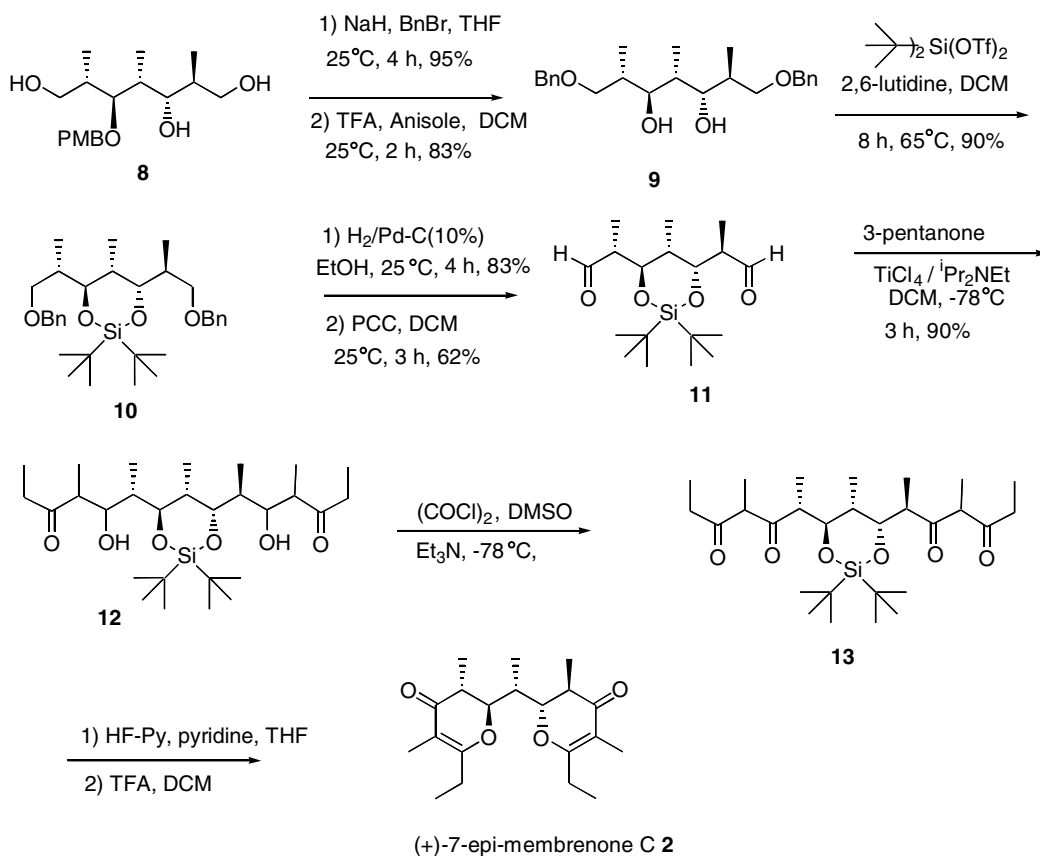
Scheme 1. Retrosynthetic strategy for (+)-membrenone C.



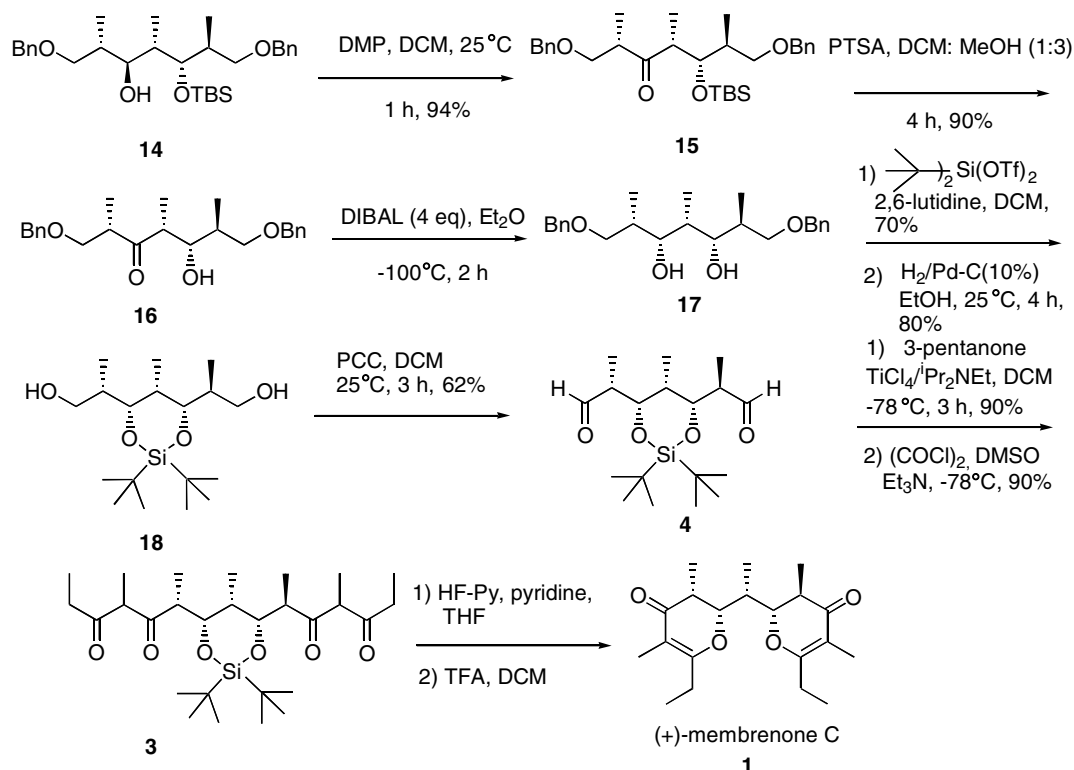
Scheme 2.

and 2,6-lutidine. Debenzylation of **10** using a catalytic amount of Pd/C and H₂ afforded a diol which on

PCC oxidation gave dialdehyde **11**. Next, the bidirectional chain extending double aldol reaction was



Scheme 3.



Scheme 4.

achieved by reacting with the Ti(IV) enolate of 3-pentanone in DCM at -78°C , as previously reported by Perkins et al. and Evans et al.^{9,10} to yield compound **12** with high diastereoselectivity (Scheme 3). Since the four stereocentres produced in the formation of **12** are not present in the final product, we proceeded with the diastereomeric mixture.

Swern oxidation of **12** gave a quantitative yield of tetraketone **13**. Since the tetraketone was very sensitive having readily epimerisable centres, we proceeded with the crude product. The di-*tert*-butyl silyl group of **13** was removed by treatment with HF·Py buffered with excess pyridine, to give a mixture of diols and hemiacetals. Rapid acid catalyzed cyclization/dehydration gave a single product **2** in 52% overall yield in three steps from **12** (Scheme 3). The synthetic material exhibited the desired spectral properties of (+)-7-*epi*-membrenone C and a specific rotation of $[\alpha]_{\text{D}}^{25} +110.6$ (c 1.81, CHCl_3).

The synthesis of (+)-membrenone C **1**, our target molecule was achieved directly from **8** which on di-*O*-benzyl protection of the primary alcohols followed by TBS protection of the secondary alcohol and deprotection of the PMB ether using DDQ in aqueous DCM¹¹ gave compound **14** in 90% overall yield in three steps. Dess–Martin oxidation (DMP)¹² of the secondary alcohol in **14** gave the ketone **15** which was subjected to desilylation using 2 equiv of PTSA in methanolic DCM¹³ to afford the hydroxy-ketone **16**. Our subsequent task was the 1,3-*syn*-selective reduction of **16** which was achieved by using DIBAL-H in diethyl ether¹⁴ at -100°C to give the desired product **17** as the major diastereoisomer with

high diastereoselectivity ($>90\%$ ds). The protection of diol **17** as the di-*tert*-butyl silylene¹⁵ followed by debenzoylation gave the diol **18** (Scheme 4). PCC oxidation of **18** gave the compound **4** which on double aldol chain extension followed by Swern oxidation gave compound **3**. Deprotection of the di-*tert*-butyl silylene group followed by cyclization/dehydration then gave **1** in 35% overall yield from **3** in two steps (Scheme 4).

The synthetic material exhibited spectral properties¹⁶ and a specific rotation, $[\alpha]_{\text{D}}^{25} +23.6$ (c 0.38, CHCl_3) {lit.³ $[\alpha]_{\text{D}}^{25} +23.5$ (c 0.63, CHCl_3)}, in agreement with those reported earlier by the Marshall group.²

In summary, the highly stereoselective synthesis of (+)-membrenone C and its 7-epimer illustrates the utility of the bicyclic precursor **6** and the desymmetrization approach control for the five required stereocentres. This is the first report of the total synthesis of (+)-7-*epi*-membrenone C.

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References and notes

- Ciavatta, L. M.; Trivellone, E.; Villani, G.; Cimino, G. *Tetrahedron Lett.* **1993**, *34*, 6791–6794.
- Marshall, J. A.; Elic, K. C. *Org. Lett.* **2003**, *5*, 1729–1732.

3. (a) Sampson, R. A.; Perkins, M. V. *Org. Lett.* **2002**, *4*, 1655–1658; (b) Sampson, R. A.; Perkins, M. V.; Taylor, M. R. *Acta Crystallogr.* **2003**, *E59*, 1867–1868; (c) Perkins, M. V.; Sampson, R. A. *Org. Lett.* **2001**, *3*, 123–126; (d) Perkins, M. V.; Sampson, R. A. *Tetrahedron Lett.* **1998**, *39*, 8367–8370.
4. Yadav, J. S.; Rao, C. S.; Chandrasekhar, S.; Ramarao, A. V. *Tetrahedron Lett.* **1995**, *36*, 7717–7720.
5. (a) Yadav, J. S.; Abraham, S.; Reddy, M. M.; Sabitha, G.; Sankar, A. R.; Kunwar, A. C. *Tetrahedron Lett.* **2001**, *42*, 4713–4716; (b) Yadav, J. S.; Abraham, S.; Reddy, M. M.; Sabitha, G.; Sankar, A. R.; Kunwar, A. C. *Tetrahedron Lett.* **2002**, *43*, 3453.
6. Yadav, J. S.; Md. Ahmed, M. *Tetrahedron Lett.* **2002**, *43*, 7147–7150.
7. Yadav, J. S.; Reddy, K. B.; Sabitha, G. *Tetrahedron Lett.* **2004**, *45*, 6475–6476.
8. Mederious, E. F. D.; Herbert, J. M.; Taylor, R. J. K. *J. Chem. Soc., Perkin Trans. 1* **1991**, 2725–2730.
9. (a) Evans, D. A.; Clark, J. S.; Metternich, R.; Novak, V. J.; Sheppard, G. S. *J. Am. Chem. Soc.* **1990**, *112*, 866–868; (b) Evans, D. A.; Urpi, F.; Somers, T. C.; Clark, J. S.; Bilodeau, M. T. *J. Am. Chem. Soc.* **1990**, *112*, 8215–8216.
10. (a) Evans, D. A.; Riegler, D. L.; Bilodeau, M. T.; Urpi, F. *J. Am. Chem. Soc.* **1991**, *113*, 1047–1049; (b) Sampson, R. A.; Perkins, M. V. *Org. Lett.* **2002**, *4*, 1655–1658.
11. Horita, K.; Yoshioka, T.; Tanaka, T.; Oikawa, Y.; Yonemitsu, O. *Tetrahedron* **1986**, *42*, 3021–3028.
12. Dess, D. B.; Martin, J. C. *J. Am. Chem. Soc.* **1991**, *113*, 7277–7287.
13. Thomas, E. J.; Williams, A. C. *J. Chem. Soc., Chem. Commun.* **1987**, 992–994.
14. Kiyooka, S.; Kuroda, H.; Shimasaki, Y. *Tetrahedron Lett.* **1986**, *27*, 3009–3012.
15. (a) Sampson, R. A.; Perkins, M. V. *Org. Lett.* **2002**, *4*, 1655–1658; (b) Trost, B. M.; Caldwell, C. G. *Tetrahedron Lett.* **1981**, *22*, 4999–5002; (c) Corey, E. J.; Hopkins, P. B. *Tetrahedron Lett.* **1982**, *23*, 4871–4874; (d) Trost, B. M.; Caldwell, C. G.; Murayama, E.; Heissler, D. *J. Org. Chem.* **1983**, *48*, 3252–3265.
16. *Spectral data of compound 1*: Low melting solid, ^1H NMR (300 MHz, CDCl_3): δ 4.24 (dd, 1H, $J = 10.4, 2.4$ Hz), 3.90 (dd, 1H, $J = 13.6, 2.4$ Hz), 2.51 (m, 1H), 2.46–2.22 (m, 5H), 2.19 (m, 1H), 1.71 (s, 3H), 1.69 (s, 3H), 1.19 (d, 3H, $J = 7.5$ Hz), 1.18 (t, 3H, $J = 8.0$ Hz), 1.11 (d, 3H, $J = 8.0$ Hz), 1.07 (t, 3H, $J = 7.2$ Hz), 1.02 (d, 3H, $J = 7.2$ Hz); ^{13}C NMR (50 MHz, CDCl_3): δ 197.0, 194.3, 173.5, 172.3, 108.4, 107.5, 83.2, 81.6, 80.8, 40.2, 39.8, 34.5, 25.3, 10.8, 10.7, 9.6, 9.4, 9.2, 9.1, 9.0; IR (Neat): 1721, 1662, 1615 cm^{-1} ; FAB Mass: m/z 335 $[\text{M}+\text{H}]^+$, $[\alpha]_{\text{D}}^{25} +23.6$ (c 0.38, CHCl_3); *Spectral data of compound 2*: liquid, ^1H NMR (300 MHz, CDCl_3): δ 4.22 (dd, 1H, $J = 8.6, 4.3$ Hz), 4.13 (dd, 1H, $J = 12.16, 1.43$ Hz), 2.48 (m, 1H), 2.42–2.34 (m, 5H), 2.18 (m, 1H), 1.75 (s, 3H), 1.71 (s, 3H), 1.25 (d, 3H, $J = 7.2$ Hz), 1.16 (t, 3H, $J = 7.8$ Hz), 1.13 (t, 3H, $J = 7.2$ Hz), 1.01 (d, 3H, $J = 7.2$ Hz), 1.00 (d, 3H, $J = 7.2$ Hz); ^{13}C NMR (75 MHz, CDCl_3): δ 195.2, 194.8, 172.6, 170.4, 108.2, 107.5, 84.3, 82.5, 80.8, 40.7, 40.0, 34.0, 25.4, 15.4, 10.9, 10.7, 9.5, 9.2, 9.1, 8.8; IR (Neat) : 1739, 1658, 1615 cm^{-1} ; FAB Mass: m/z 335 $[\text{M}+\text{H}]^+$, HRMS calcd for $\text{C}_{20}\text{H}_{30}\text{O}_4$: 335.2222. Found: 335.2251. $[\alpha]_{\text{D}}^{25} +110.6$ (c 1.81, CHCl_3).